

LETTERS TO THE EDITOR

Synthetic Approach to β -Cyclodextrin Derivatives

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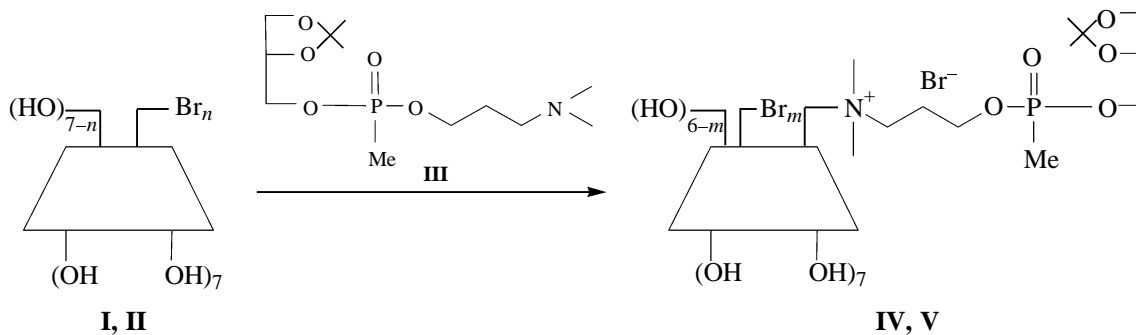
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Previously we showed [1] a fundamental possibility of preparing 6-bromo-6-deoxy- β -cyclodextrins containing a preset number of bromine atoms ($n = 1$ –7). Due to their alkylating ability toward organic amines, these derivatives attract special attention as starting materials for synthesis of alkylammonium (cationic) cyclodextrin derivatives that present some practical interest [2, 3]. Here we report on the synthesis of cationic derivatives of β -cyclodextrins containing a 1,2-*O,O*-isopropylideneglyceromethylphosphonate residue from 6-bromo-6-deoxy- β -cyclo-

dextrins with an average bromine content (n) of $n = 1$ (compound **I**) and $n = 2.9$ (compound **II**). Such compounds are important intermediates for subsequent synthesis of complex glycerophospholipid analogs containing a cationic β -cyclodextrin fragment.

To this end, we reacted β -cyclodextrin bromides **I** and **II** [1] with 3 equiv of methylphosphonate **III** prepared from methylphosphonic dichloride, 1,2-*O,O*-isopropylideneglycerol, and 3-(*N,N*-dimethylamino)propan-1-ol.



$n = 1$ (**I**), 2.9 (**II**); $m = 0$ (**IV**), 1.9 (**V**).

It is important that under these conditions only 1 equiv of compound **III** was alkylated to form compounds **IV** and **V**. At the same time, no reaction between equimolar amounts of bromo derivative **II** and methylphosphonate **III** took place under the same conditions probably because the bromine atom in derivative **II** is shielded and (or) reagent **III** is incorporated into the cyclodextrin cavity. Note that an analogous specific supramolecular effect of the cyclodextrin cavity on reaction pathway is a characteristic feature of cyclodextrin and their derivatives [2, 4].

(6-Bromo) $_m$ -(6-deoxy) $_m$ -6-{3-[2,3-(isopropylidenedioxy)propoxy]methylphosphoryloxy}propyl}-dimethylammonio- β -cyclodextrin bromide (IV**), ($m = 0$).**

To a suspension of 0.1 g of bromodeoxy derivative **I** in 2 ml of DMF, 0.074 g of 3-(*N,N*-dimethylamino)propyl 2,3-(isopropylidenedioxy)propyl methylphosphonate (**III**) in 0.7 ml of DMF was added. The reaction mixture was stirred for 96 h at 20°C and then diluted with 5 ml of diethyl ether. The precipitate obtained was filtered off, washed with diethyl ether, and dried in a vacuum (1 mm) for 9 h at 50°C. Yield

0.0923 g (74%), mp 232–234°C (decomp.). ^1H NMR spectrum ($\text{DMSO}-d_6$), δ , ppm (J , Hz): 1.28 s (3H, OCCH_3), 1.32 s (3H, OCCH_3), 1.46 d (3H, PCH_3 , $^2J_{\text{PH}}$ 17.26), 1.79 m (2H, $\text{CH}_2\text{CH}_2\text{N}$), 2.71 m (2H, CH_2N), 3.29 s (6H, NCH_3), 3.33–3.63 m (42H, $\text{C}^2\text{H}-\text{C}^5\text{H}$, C^6H_2), 3.98 m (2H, CHCH_2C), 4.00 m (2H, POCH_2CH_2), 4.23 m (2H, CHCH_2OP), 4.36 m (1H, CH_2CHCH_2), 4.42 br.s (6H, C^6OH), 4.82 br.s (7H, C^1H), 5.73 m (14H, C^2OH , C^3OH). ^{31}P NMR spectrum (DMF), δ_{P} , ppm: 30.5. Found, %: C 43.50; H 6.29; P 2.10. $\text{C}_{54}\text{H}_{95}\text{BrNO}_{39}\text{P}$. Calculated, %: C 43.44; H 6.41; P 2.07.

(6-Bromo) $_m$ -(6-deoxy) $_m$ -6-[3-[2,3-(isopropylidenedioxy)propoxy]methylphosphoryloxy]propyl}-dimethylammonio- β -cyclodextrin bromide (V) ($m = 1.9$). Compound V ($m = 1.9$) was prepared analogously to compound IV from 0.1 g of bromodeoxy derivative II and 0.065 g of methylphosphonate III. Yield 0.0876 g (72%), mp 209–210°C (decomp.). ^1H NMR spectrum ($\text{DMSO}-d_6$), δ , ppm (J , Hz): 1.29 s (3H, OCCH_3), 1.34 s (3H, OCCH_3), 1.47 d (3H, PCH_3 , $^2J_{\text{PH}}$ 17.26), 1.91 m (2H, $\text{CH}_2\text{CH}_2\text{N}$), 2.81 m (2H, CH_2N), 3.21 s (6H, NCH_3), 3.31–3.64 m (42H, $\text{C}^2\text{H}-\text{C}^5\text{H}$, C^6H_2), 3.94 m (2H, CHCH_2C), 4.00 m (2H, POCH_2CH_2), 4.23 m (2H, CHCH_2OP), 4.36 m (1H, CH_2CHCH_2), 4.41 br.s (4H, C^6OH), 4.85 br.s (7H, C^1H), 5.6 m (14H, C^2OH , C^3OH). ^{31}P NMR spectrum (DMF), δ_{P} , ppm: 30.4. Found, %: C 40.10; H 5.34; P 1.88. $\text{C}_{54}\text{H}_{93}\text{Br}_{2.9}\text{NO}_{37}\text{P}$. Calculated, %: C 40.06; H 5.79; P 1.91.

The ^1H NMR spectra were taken on a Bruker AC-200 spectrometer (200 MHz), internal reference TMS. The $^{31}\text{P}-\{^1\text{H}\}$ NMR spectra were registered on a Bruker WR-80SY spectrometer (32.4 MHz), external reference 85% phosphoric acid.

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REFERENCES

1. Kurochkina, G.I., Trushkin, I.Yu., Gratchev, M.K., and Nifant'ev, E.E., *Zh. Obshch. Khim.*, 2004, vol. 74, no. 10, p. 1743.
2. Khan, A.R., Forgo, P., Stine, K.J., and D'Souza, V.T., *Chem. Rev.*, 1998, vol. 98, no. 5, p. 1977.
3. Gratchev, M.K., Mustafin, I.G., and Nifant'ev, E.E., *Zh. Obshch. Khim.*, 1998, vol. 68, no. 9, p. 1519.
4. Glazyrin, A.E., Kurochkina, G.I., Gratchev, M.K., and Nifant'ev, E.E., *Mendeleev Commun.*, 2001, no. 6, p. 218; Glazyrin, A.E., Syrtsev, A.N., Kurochkina, G.I., Kononov, L.O., Gratchev, M.K., and Nifant'ev, E.E., *Izv. Ross. Akad. Nauk, Ser. Khim.*, 2003, no 1, p. 225.